

- ① Electronic transition
② Rare earths
③ Thermal expansion

Reprinted from
Journal of the Less-Common Metals
Elsevier Publishing Company, Amsterdam
Printed in The Netherlands

K.J. 74

GSCH KA65 0223

THERMAL EXPANSION OF LaRu₂, CeRu₂ AND PrRu₂ FROM 20° TO 900°C*

K. A. GSCHNEIDNER, JR.,** R. O. ELLIOTT AND D. T. CROMER

University of California, Los Alamos Scientific Laboratory, Los Alamos, N.M. (U.S.A.)

(Received November 7th, 1964)

SUMMARY

The lattice parameters of LaRu₂, CeRu₂ and PrRu₂ have been measured from approximately 20° to 800°, 900° and 650°C, respectively. The lattice parameters expand smoothly and continuously as the temperature is increased. CeRu₂, however, expands at a much greater rate than either LaRu₂ or PrRu₂. This greater rate is thought to be due to the transfer of 0.07 electron from the valence band to a one-electron 4*f* band by thermal excitation.

INTRODUCTION

Cerium metal is known to undergo an electronic transformation*** at either 1 atm and 116°K or 7,670 atm and 298°K^{1,2}. This transition involves the transfer of about 0.6 electron per atom from the 4*f* band to the 6*s5d* valence band as the cerium is cooled or compressed². For a number of intermetallic compounds cerium is known to be in the higher valent form at standard temperature and pressure. Thus, it was thought that heating one of these compounds might cause the cerium to undergo an electronic transformation from the tetravalent to the trivalent state. If a compound should undergo such a transformation then one would expect a large volume change to occur, which could be easily detected by thermal expansion measurements.

CeRu₂ was chosen as the compound to be studied because it has a cubic structure (C15, MgCu₂ type Laves phase) and because its lattice constant deviates farther from the smooth curve established for the corresponding trivalent RRu₂ Laves phase compounds (where R = La, Pr, Nd etc.) than do those of any of the other CeM₂ compounds (where M = Mg, Al, Ni, Rh, Os, Ir and Pt) from the respective curves established for the RM₂ compounds. GSCHNEIDNER³ shows that for CeM₂ compounds the difference between the theoretical lattice constants (for trivalent cerium only) and

* Work performed under the auspices of the U.S. Atomic Energy Commission.

** Present address: Department of Metallurgy and Institute for Atomic Research, Iowa State University, Ames, Iowa, (U.S.A.).

*** Although this electronic transformation has been known for nearly 15 years the reader is referred to two recent papers (refs. 1 and 2), which summarize the older pertinent literature and also more recent data.